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Biaryl Synthesis from Two Different Aryl Halides with Tri(2-furyl)germane

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ABSTRACT

The coupling reaction of germanium compounds with aryl halides has been developed. The Pd(0)-catalyzed reaction of aryl halides with tri(2-furyl)germane provides aryltri(2-furyl)germanes in good yield. The cross-coupling reaction of aryltri(2-furyl)germanes with aryl halides is acheived. This allows facile synthesis of unsymmetrical biaryls from two different aryl halides.

Group 14 metals are popular key elements among a number of cross-coupling reactions between organometallic compounds and organic halides under Pd(0) catalysis (Scheme 1). The Stille coupling is particularly important because of

Scheme 1. Coupling Reactions with Group 14 Metals

Ar¹—MR₃ + Ar²-X Pd(0) Ar¹-Ar²

M = Si: Hiyama Coupling

M = Ge: ?

M = Sn: Stille Coupling

its mild reaction conditions and wide functional compatibility.² Since Denmark and Mori independently have found that coupling with silanols proceeds under milder conditions than Hiyama's original procedure,³ the Hiyama coupling has

received increasing attention due to the stability and low toxicity of silicon compounds.⁴ However, despite the wide application of these two coupling methods, coupling with organogermanes has been scarcely explored so far, although germanium is located between silicon and tin in a periodic table.⁵

Recently, we have investigated unique properties of tri-(2-furyl)germyl compounds.⁶ We have also devoted our effort to develop an efficient coupling reaction with arylgermanes to bridge the apparent gap between arylsilanes and arylstannanes in the cross-coupling chemistry. Herein, we wish to report the cross-coupling reactions between aryltri(2-furyl)germanes and aromatic halide in the presence of a Pd(0) catalyst. Aryltri(2-furyl)germanes are readily accessible from aryl halides and tri(2-furyl)germane, and this allows facile synthesis of unsymmetrical biaryls from two different aryl halides.

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Table 1. Palladium-Catalyzed Germylation of Aryl or Vinyl Halides^a

$$(\bigcirc)_3 \text{GeH} + \text{RX} \quad \frac{\text{Pd(OAc)}_2, \text{DPPF, Cs}_2\text{CO}_3}{\text{DMF, r. t.}} \quad (\bigcirc)_3 \text{GeR}$$

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entry	RX	time (h)	product	yield (%)
1	Phi	1.5	2a	83
2	4-MeOC ₆ H ₄ I	1	2b	80
3	3-MeOC ₆ H ₄ I	1	2c	88
4	4-MeC ₆ H ₄ I	0.5	2d	85
5	2-MeC ₆ H ₄ I	2	2e	43
6	4-M e ₂ NC ₆ H ₄ I	1	2f	56
7	n-C ₅ H ₁₁ n-C ₅ H ₁₁	0.5	2g	83
8	ОН	3	2h	59
9	4-MeOC ₆ H ₄ Br	12	_	0
10	Br	1.5	2i	70
11	Br	1	2j	54
12	α -Bromostyren e	1	2k	49
13	(E)-β-Bromostyrene	1.5	21	74
14	2-Bromopyridine	2	2m	73

 a Tri(2-furyl)germane (1.0 mmol), RX (1.2 mmol), Pd(OAc) $_2$ (0.050 mmol), DPPF (0.050 mmol), Cs $_2$ CO $_3$ (1.1 mmol), and DMF (10 mL) were employed.

We examined the coupling reaction of tri(2-furyl)germane (1) with a variety of aryl iodides. ⁷ To a mixture of Pd₂(dba)₃• CHCl₃ (0.025 mmol), tri(2-furyl)phosphine (0.10 mmol), and Cs₂CO₃ (1.1 mmol) in DMF were sequentially added tri(2furyl)germane (1) (1.0 mmol) and iodobenzene (1.2 mmol) at room temperature. The mixture was stirred for 1 h. Aqueous workup and purification afforded arylgermane 2a in 61% yield. After extensive optimization of the reaction conditions, we found that a Pd(OAc)₂/DPPF/Cs₂CO₃ combination was particularly effective and improved the yield of 2a up to 83%. In the absence of a ligand or a base, the desired product was not obtained at all. It is striking that the reaction of 4-iodoanisole provided a trace amount of anisole, while the analogous reaction with Et₃SiH⁸ or *n*-Bu₃SnH⁹ resulted in the formation of anisole via the reduction of halide.

Table 1 summarizes the results of germylation of aryl and alkenyl halides under the optimized reaction conditions. Electron-rich aryl iodides such as iodoanisoles and 4-iodo-

toluene afforded the corresponding arylgermanes in good yields (Table 1, entries 2–4). The use of 2-iodotoluene lowered the yield of the coupling product probably due to the steric hindrance at the ortho position (Table 1, entry 5). Not only aryl iodides but also vinyl halides can be employed (Table 1, entries 7 and 8). Disappointingly, aryl bromide did not undergo coupling (Table 1, entry 9). In contrast, vinyl bromides and 2-bromopyridine furnished the corresponding products in moderate to good yields (Table 1, entries 10–14).

We then turned our attention toward the cross-coupling reaction of aryltri(2-furyl)germanes with aryl halides. After extensive investigation of the reaction conditions, we found that an addition of fluoride induced the cross-coupling of 4-methoxyphenylgermane **2b** with aryl halides (Table 2). The

Table 2. Cross-Coupling Reaction with Iodobenzene^a

TBAF

OMe + Ph-I
$$\frac{Pd, (2-\text{furyI})_3P}{\text{NMP, } 100 °C}$$
 Ph

2b $Ge = (2-\text{furyI})_3 \text{ Ge}$ 3a

entry	Pd	TBAF ^b (equiv)	time (h)	yield (%)
1	Pd(OAc) ₂	1.2	5	0
2	Pd(OAc) ₂	2.4	10	23
3	Pd(OAc) ₂	3	5	52
4	Pd(OAc) ₂	4	5	77
5	Pd(OAc) ₂	5	5	81
6^c	Pd(OAc) ₂	5	10	0
7	$Pd_2(dba)_3 \cdot CHCl_3$	4	5	90

^a Arylgermane **2b** (1.0 mmol), PhI (1.2 mmol), Pd(OAc)₂ (0.020 mmol), or Pd₂(dba)₃•CHCl₃ (0.010 mmol), (2-furyl)₃P (0.020 mmol), and NMP (10 mL) were employed. ^b 1.0 M THF solution. ^c The reaction was carried out at 60 °C.

amount of TBAF was crucial for the successful coupling. To obtain the desired biaryl in satisfactory yield, more than 4.0 equiv of TBAF was required (Table 2, entries 1–5). As the palladium source, Pd₂(dba)₃•CHCl₃ proved to be the best (Table 2, entry 7).

The key of this coupling process is the use of the tri(2-furyl)germyl group. It is noteworthy that none of the coupling reaction proceeded with Ph₄Ge under numerous reaction conditions with or without fluoride. To obtain mechanistic insights of this process, we examined the reaction of an aryltri(2-furyl)germane with TBAF. Arylgermane **2b** was treated with 4.0 equiv of TBAF in THF at reflux for 5 h. After solvent was removed, ¹H and ¹³C NMR spectra of the residue indicated the presence of a 4-methoxyphenyl group but not furyl groups. In the ⁹F NMR, no other peaks than TBAF were detected. These observations indicate that all three furyl groups on germanium were cleaved with fluoride to form ArGe(OH)₃ or germanoxanes. ¹⁰ We speculate that

3166 Org. Lett., Vol. 4, No. 18, 2002

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Table 3. Cross-Coupling of Aryltri(2-furyl)germane with Aryl Halides^a

entry	$ m Ar^1$	$\mathrm{Ar}^2\mathrm{X}$	time (h)	product	yield (%)
1	4-MeOC ₆ H ₄ 2b	PhI	5	3a	90
2	2b	PhBr	6.5	3a	100
3	2b	$3-CF_3C_6H_4Br$	1	3b	80
4	2b	1-iodonaphtalene	4	3c	70
5	2b	3-MeOC_6H_4I	2	3d	59
6	2b	4-MeC_6H_4I	1	3e	56
7	Ph 2a	PhI	5	3f	61
8	2a	$3-CF_3C_6H_4Br$	9	3g	64
9	2a	1-iodonaphtalene	6	3h	52
10	3-MeOC ₆ H ₄ 2c	PhI	5	3i	59
11	2c	$3-CF_3C_6H_4Br$	10	3j	64
12	2c	1-iodonaphtalene	10	3k	65
13	(<i>E</i>)−1-Octenyl 2n	PhI	8	31	60

 $[^]a$ Arylgermane 2 (1.0 mmol), Ar²X (1.2 mmol), Pd₂(dba)₃·CHCl₃ (0.010 mmol), (2-furyl)₃P (0.020 mmol), TBAF (4.0 mmol), and NMP (10 mL) were employed.

nucleophilic hypervalent oganogermanium species such as [ArGe(OH)₃F]⁻ undergo transmetalation from germanium to palladium.

Table 3 summarizes the results of the cross-coupling reaction with a variety of aryl halides. 11 Aryl bromides as

well as aryl iodides afforded the desired products in good yields (Table 3, entries 1 and 2). Electron-donating groups on aryl halides slightly lowered the yields of biaryls (Table 3, entries 5 and 6). Treatment of aryl or vinyl germanes (2a, 2c, and 2n) with aryl halides yielded the corresponding biaryls or styrenes in moderate yields (Table 3, entries 7–13). In all cases, none of furylated products were observed.

In summary, we have achieved the coupling reaction between tri(2-furyl)germane and aryl halides providing aryltri(2-furyl)germanes, which can be further cross-coupled with a variety of halides under Pd(0) catalysis. This new protocol enables the facile synthesis of unsymmetrical biaryls from two different aryl halides.

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Supporting Information Available: General procedures and spectral data for compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Org. Lett., Vol. 4, No. 18, 2002

⁽¹¹⁾ We have attempted the synthesis of unsymmetrical biaryls in a one-pot operation. Tri(2-furyl)germane was treated with 4-iodoanisole in the presence of Pd(OAc)₂, DPPF, and Cs₂CO₃ in DMF. After 1 h, iodobenzene, TBAF, and NMP were added, and then the mixture was heated at 100 °C for 10 h. 4-Methoxybiphenyl $\bf 3a$ was obtained in 34% yield.